Tetrahedron Letters 51 (2010) 4538-4542

Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

Highly regioselective synthesis of glycospiro heterocycles through 1,3-dipolar cycloaddition reaction

R. Prasanna, S. Purushothaman, R. Raghunathan *

Department of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600 025, India

ARTICLE INFO

Article history: Received 8 April 2010 Revised 16 June 2010 Accepted 21 June 2010 Available online 25 June 2010

Keywords: Glyco-heterocycles Azomethine ylides 1,3-DC reaction Spiropyrrolidine

ABSTRACT

A highly regio-selective synthesis of novel glycospiropyrrolidines has been accomplished by 1,3-dipolar cycloaddition (1,3-DC) reaction. A unique dipolarophile derived from galactose has been reacted with azomethine ylide generated from 1,2-diketones and secondary aminoacids to give the corresponding spiro glycoheterocycles in good yields. The structures were assigned by 2D NMR spectra and the regioand stereochemical outcome of the cycloadducts was established by a single crystal X-ray analysis. © 2010 Elsevier Ltd. All rights reserved.

The surge of interest for carbohydrate-derived heterocycles¹ is due to their potential biological significance and pharmaceutical applications.² Recently, several studies have demonstrated that the carbohydrate-based heterocycles have shown a diverse range of bioactivities such as anti-influenza (H1N1),³ antitumor,⁴ anti-HIV,⁵ antimicrobial,⁶ and anticancer,⁷ thereby making these compounds a particularly important subject for study. The diversity of biological activity and the ever-growing new application of these compounds have stimulated a great deal of interest in regard to their synthesis.

1,3-DC reaction of azomethine ylide is a powerful tool for the synthesis of a variety of natural products⁸ containing a pyrrolidine structure. Although many investigations have been made to exploit carbohydrate as a template for 1,3-DC reaction involving azide,⁹ nitrone,¹⁰ and nitrile oxide,¹¹ there are only very few reports involving azomethine ylide¹² in the literature and the synthetic utility of azomethine ylide reaction has not been well exploited in carbohydrate-fused heterocyclic systems through 1,3-DC reaction using azomethine ylide as a dipole.

In continuation of our research in the field of 1,3-dipolar cycloaddition reaction,¹³ herein we report a highly efficient protocol for the synthesis of sugar-derived spiropyrrolidines through 1,3-DC reaction from D-galactose. 1,2:3,4-diisopropylidene-D-galactopyranose¹⁴ on oxidation¹⁵ gave the corresponding aldehyde **1** which on Wittig olefination¹⁶ with the phosphorus ylide **2** derived from ethylbromoester gave the conjugated ester **3** in a 80:20 diastereomeric ratio with the predominant formation of *E* isomer (Scheme 1). These diastereomers were separated by column chromatography to give pure compounds **3a** and **3b** in good yield. The structure and geometry of the olefinic ester **3a** and **3b** were deduced on the basis of ¹H NMR spectral data where the coupling constant for **3a** (J = 15.9 Hz) and **3b** (J = 11.7 Hz) of vinylic protons confirmed a *trans* (*E*) and *cis* (*Z*) geometry of the products, respectively.

etrahedro

These dipolarophiles **3a** and **3b** when reacted with azomethine ylide generated from acenaphthequinone **4a** and sarcosine **5a** in refluxing toluene under Dean–Stark reaction conditions, furnished the cycloadducts¹⁷ **6a** and **6b** in good yields (88–92%) in a highly regioselective and stereoselective manner¹⁸ through an intermolecular 1,3-dipolar cycloaddition reaction (Scheme 2).

We then studied the solvent effects on the cycloaddition reaction for the synthesis of **6a** and **6b**. Interestingly, the reaction pro-



Scheme 1. Synthesis of glycoacrylate 3a-b.



^{*} Corresponding author. Tel.: +91 44 22202811; fax: +91 44 22300488. *E-mail address:* ragharaghunathan@yahoo.com (R. Raghunathan).

^{0040-4039/\$ -} see front matter \odot 2010 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2010.06.098



Scheme 2. Synthesis of glycopyrrolidine.

Table	1		
Effect	of solvent of	on the yiel	d of reaction

Solvent	Temperature (°C)	Time		Yield ^a (%)	
		6a	6b	6a	6b
Toluene	Reflux	4	5	92	88
Benzene	Reflux	8	8	57	52
CH ₃ CN	Reflux	6	6	76	78
CH₃OH	Reflux	10	12	30	34
DMF	110	12	12	16	12
1,4-Dioxane	90	8	8	18	12

^a Isolated yield in percentage.



Selective HMBC correlations of 6a

Figure 1. HMBC correlation of 6a.

Table 2 Synthesis of galactopyranoyl-derived spiroheterocycles through [3+2] cycloaddition reaction

Entry	Product ^a	Time (h)	Yield ^b (%)
1	7a	4.5	86
2	8a	5.0	78
3	7b	4.0	82
4	8b	5.0	76
5	9a	4.0	84
6	9b	4.5	86
7	10a	5.5	82
8	11a	5.5	84
9	10b	4.5	88
10	11b	5.0	86

^a Reaction carried out with toluene/reflux.

^b Isolated yields of pure products.

ceeded more efficiently in toluene than in benzene or other solvents (Table 1). The structures of the products were confirmed by IR, ¹H NMR, ¹³C NMR, DEPT 135, HMBC, and HRMS.

The ¹H NMR spectrum of diastereomers **6a** and **6b** showed a doublet at δ 3.60 (*J* = 9.6 Hz) and 3.24 (*J* = 6.9 Hz), respectively, for the H_a proton, which clearly showed the stereo- and regiochemistry of the cycloaddition reaction.

In the H–H COSY spectrum of **6a**, the proton at δ 3.60 (H_a) showed a correlation with the proton at δ 3.26 (H_b), whereas the peak at δ 3.26 (H_b) showed a H–H COSY correlation with the peak at δ 3.95 (Gal-C₅-H) and the peaks at δ 3.15 and δ 3.65 (diastereotropic –CH₂–), which clearly indicated that the signals at δ 3.60 are



Scheme 3. Synthesis of glycopyrrolizidines.



Scheme 4. Synthesis of glycopyrrolizidines.



Scheme 5. Synthesis of spirooxindolopyrrolizidines.



Figure 2. ORTEP diagram of 9b.

due to the H_a proton and δ 3.26 is due to the H_b proton. The doublet at δ 3.60 (*J* = 9.6 Hz) and the multiplet at δ 3.26 have an HMBC contour (Fig. 1) with carbons at 169.8 ppm (ester carbonyl carbon) and 75.5 ppm (spiro carbon) which further confirmed the δ values of the H_a and H_b protons, respectively.

Furthermore, the presence of a molecular ion peak at m/z 537 (M⁺) in the mass spectrum confirmed the structure of the cycloadduct **6a**.

The same reaction was carried out with cyclic amino acid L-proline **5b** and thiazolidine-4-carboxylic acid **5c** in order to obtain a



Figure 3. ORTEP diagram of 10b.

glycospiropyrrolizidines and pyrrolothiazoles,¹⁹ respectively (Scheme 3, Table 2). The structures of the products (**7** and **8**) were confirmed by spectroscopic techniques.

In order to further extend the scope of the reaction, the dipole generated from isatin **4b** (a diketone) and the secondary amino acids **5a–c** reacted with **3a** and **3b** to give novel sugar-derived spirooxindolopyrrolidines (Scheme 5), pyrrolizidines, and pyrrolothiazoles in good yields,²⁰ respectively (Scheme 4 and 5, Table 2). The structures of the products (**9–11**) were also confirmed by spectroscopic techniques. The ¹H NMR spectrum of **9a** exhibited a doublet at δ 3.57 (*J* = 9.0 Hz) for the H_a proton, which clearly showed the regio- and stereo chemistry of the cycloaddition reaction.

Finally, the regio- and stereochemical outcome of the cycloaddition reaction was confirmed by a single crystal X-ray analysis²¹ of the cycloadducts **9b** (Fig. 2) and **10b** (Fig. 3).

In conclusion, we have developed a simple and efficient protocol for the synthesis of sugar-derived heterocycles containing pyrrolidines, pyrrolizidines, and pyrrolothiozoles through 1,3-dipolar cycloaddition methodology. Further work in this direction is in progress.

Acknowledgments

The authors R.P. and R.R. thank DST (Project No. SR/S1/OC-51/2008), New Delhi for the fellowship and DST-FIST program for the NMR facility. The authors wish to thank Professor D. Velmurugan and S. Sundaramoorthy for single crystal X-ray data. S.P. thanks CSIR New Delhi for the award of SRF.

References and notes

- Doddi, V. R.; Kokatla, H. P.; Pal, A. P. J.; Basak, R. K.; Vankar, Y. D. Eur. J. Org. Chem. 2008, 5731–5739.
- Calarese, D. A.; Scanlan, C. N.; Zwick, M. B.; Deechongkit, S.; Mimura, Y.; Kunert, R.; Zhu, P.; Wormald, M. R.; Stanfield, R. L.; Roux, K. H.; Kelly, J. W.; Rudd, P. M.;

Dwek, R. A.; Katinger, H.; Burton, D. R.; Wilson, I. A. Science **2003**, 300, 2065–2071.

- Ferguson, N. M.; Cummings, D. A. T.; Cauchemez, S.; Fraser, C.; Riley, S.; Meeyai, A.; Iamsirithaworn, S.; Burke, D. S. *Nature* 2005, 437, 209–214.
- Hosoya, T.; Takashiro, E.; Matsumoto, T.; Suzuki, K. Tetrahedron Lett. 1994, 35, 4591–4594.
- Kim, H. O.; Ahn, S. K.; Alves, A. J.; Beach, J. W.; Jeong, L. S.; Choi, B. G.; Roey, P. V.; Schinazi, R. F.; Chu, C. K. J. Med. Chem. **1992**, 35, 1987–1995.
- (a) Cipolla, L.; La Ferla, B.; Nicotra, F. Curr. Top. Med. Chem. 2003, 3, 485–511; (b) Harris, C. M.; Harris, T. M. J. Am. Chem. Soc. 1982, 104, 363–365.
- (a) Danishefsky, S. J.; Allen, J. R. Angew. Chem., Int. Ed. 2000, 39, 836–863; (b) Allen, J. R.; Harris, C. R.; Danishefsky, S. J. J. Am. Chem. Soc. 2001, 123, 1890– 1897.
- (a) Tufariello, J. J.; Meckler, H.; Senaratne, K. P. A. *Tetrahedron* **1985**, *41*, 3447–3453;
 (b) Daly, J. W.; Spande, T. W.; Whittaker, N.; Highet, R. J.; Feigl, D.; Noshimori, N.; Tokuyama, T.; Meyers, C. W. *J. Nat. Prod.* **1986**, *46*, 265–280;
 (c) Coulter, T.; Grigg, R.; Malone, J. F.; Sridharan, V. *Tetrahedron Lett.* **1991**, *32*, 5417–5420.
- (a) Kumar, A.; Pandey, P. S. Org. Lett. 2008, 10, 165–168; (b) Kuijpers, B. H. M.; Groothuys, S.; Keereweer, A.; Bram, R.; Quaedflieg, P. J. L. M.; Blaauw, R. H.; van Delft, F. L.; Rutjes, F. P. J. T. Org. Lett. 2004, 6, 3123–3126; (c) Hotha, S.; Anegundi, R. I.; Natu, A. Tetrahedron Lett. 2005, 46, 4585–4588; (d) Koumbis, A. E.; Gallos, J. K. Curr. Org. Chem. 2003, 7, 771–797.
- (a) Socha, D.; Jurczak, M.; Chmielewski, M. Carbohydr. Res. 2001, 336, 315–318;
 (b) Panfil, I.; Solecka, J.; Chmielewski, M. J. Carbohydr. Chem. 2006, 25, 673–684;
 (c) Paśniczek, K.; Socha, D.; Solecka, J.; Jurczak, M.; Chmielewski, M. Can. J. Chem. 2006, 84, 534–539;
 (d) Bernet, B.; Vasella, A. Helv. Chim. Acta 1979, 62, 1990–2016;
 (e) Koumbis, A. E.; Gallos, J. K. Curr. Org. Chem. 2003, 7, 585–628;
 (f) Osborn, H. M. I.; Gemmell, N.; Harwood, L. M. J. Chem. Soc., Perkin Trans. 1 2002, 2419–2438;
 (g) Enderlin, G.; Taillefumier, C.; Didierjean, C.; Chapleur, Y. Tetrahedron: Asymmetry 2005, 16, 2459–2474.
- (a) Gallos, J. K.; Koftis, T. V. J. Chem. Soc., Perkin Trans. 1 2001, 415–423; (b) RajanBabu, T. V.; Reddy, G. S. J. Org. Chem. 1986, 51, 5458–5461; (c) Gallos, J. K.; Koftis, T. V.; Koumbis, A. E.; Moutsos, V. I. Synlett 1999, 1289–1291; (d) Gallos, J. K.; Koumbis, A. E. Curr. Org. Chem. 2003, 7, 397–426; (e) Benltifa, M.; Vidal, S.; Gueyrard, D.; Goekjian, P. G.; Msaddek, M.; Praly, J. P. Tetrahedron Lett. 2006, 47, 6143–6147.
- (a) Karthikeyan, K.; Kumar, R. S.; Muralidharan, D.; Perumal, P. T. *Tetrahedron* Lett. 2009, 50, 7175–7179; (b) Sirisha, N.; Raghunathan, R. *Tetrahedron Lett.* 2010, 51, 2515–2518.
- (a) Subramaniyan, G.; Raghunathan, R.; Castro, A. M. M. Tetrahedron 2003, 59, 335–340; (b) Jayagobi, M.; Mahalingam Poornachandran, M.; Raghunathan, R. Tetrahedron Lett. 2009, 50, 648–650; (c) Poornachandran, M.; Raghunathan, R. Tetrahedron: Asymmetry 2008, 19, 2177–2183; (d) Purushothaman, S.; Raghunathan, R. Tetrahedron Lett. 2009, 50, 6848–6850.
- 14. Liu, H.; Nakanishi, K. J. Am. Chem. Soc. 1982, 104, 1178-1185.

- Katiyar, D.; Mishra, R. C.; Tripathi, R. P. J. Carbohydr. Chem. 2004, 23, 493–511.
 (a) Tronchet, M. J. J.; Tronchet, J. Carbohydr. Res. 1974, 33, 237–248; (b) Harcken, C.; Martin, S. F. Org. Lett. 2001, 3, 3591–3593.
- 17. General procedure for synthesis cycloadducts (6-11): A mixture of diketone (4a-b) (1 equiv), ethyl 6,7-dideoxy-1,2:3,4-di-O-isopropylidene-α-D-galacto-6-eno-octopyranuronate (3a-b) (1 equiv), and a cyclic or acyclic secondary amino acid (5a-c) (1.2 equiv) was refluxed in dry toluene under N₂ for about 4-6 h under Dean-stark reaction condition to give cycloadduct (6-11). After the completion of reaction as indicated by TLC, the solvent was evaporated under reduced pressure. The residual mass was dissolved in chloroform (50 mL) and washed with water (50 mL). The organic layer was dried (Na₂SO₄) and evaporated under reduced pressure to give a crude mass. The crude product was purified by column chromatography using hexane/EtOAc (8:2) as an eluent. (a) Synthesis of trans-1,2:3,4-diisopropylidene-5-C[acenaphthene-1"-onespiro[2".2']-(3'-ethoxycarbonyl)-1'-N-methylpyrrolidine]-α-D-galactopyranose (6a): isolated yield: 92% (0.440 g, 0.82 mmol) from 0.30 g of (E)-ethyl 6,7-dideoxy-1,2:3,4-di-0-isopropylidene-α-D-galacto-6-eno-octopyranuronate (0.91 mmol). Yellow solid. Mp = 199 °C. ¹H NMR (300 MHz, CDCl₃) δ = 0.12 (t, the solid solid. Mp = 199 °C. ¹H NMR (300 MHz, CDCl₃) δ = 0.12 (t, the solid solid solid. Mp = 199 °C. ¹H NMR (300 MHz, CDCl₃) δ = 0.12 (t, the solid solid solid solid solid solid solid. Mp = 199 °C. ¹H NMR (300 MHz, CDCl₃) δ = 0.12 (t, the solid solid
 - (0.91 mmol). Yellow solid. Mp = 199 °C. 'H NMR (300 MHz, CDCl₃) $\delta = 0.12$ (t, J = 7.2 Hz, 3H); 1.25 (s, 3H); 1.26 (s, 3H); 1.40 (s, 3H); 1.40 (s, 3H); 1.40 (s, 3H); 1.91 (s, 3H); 1.5 (t, J = 9.3 Hz, 1H); 3.24–3.28 (m, 1H); 3.34 (q, J = 7.2 Hz, 2H); 3.60 (d, J = 9.6 Hz, 1H); 3.64–3.67 (m, 1H); 3.94–3.95 (m, 1H); 4.22 (dd, J = 1.5, 7.8 Hz, 1H); 4.24–4.27 (m, 1H); 4.56 (dd, J = 2.1, 7.8 Hz, 1H); 5.58 (d, J = 5.1 Hz, 1H); 7.43 (d, J = 6.9 Hz, 1H); 7.53 (t, J = 6.9 Hz, 1H); 7.65 (t, J = 7.2 Hz, 1H); 7.78 (d, J = 8.4 Hz, 1H); 7.89 (d, J = 7.2 Hz, 1H); 8.03 (d, J = 8.4 Hz, 1H); 7.89 (d, J = 7.2 Hz, 1H); 8.03 (d, J = 8.4 Hz, 1H); 7.89 (d, 6.9.5, 70.1, 72.1, 75.5, 95.7, 107.5, 108.2, 120.1, 121.0, 123.9, 126.9, 127.2, 129.5, 130.5, 131.6, 135.4, 141.2, 169.8, 204.6. IR (KBr) ν_{max} : 1730, 1718 cm⁻¹. HRMS (EI) exact mass calcd for $C_{30}H_{35}NO_8$: 537.2363 (M*), found 537.2364. Anal. Calcd for $C_{30}H_{35}NO_8$: C, 67.02; H, 6.56; N, 2.61. Found: C, 67.08; H, 6.65; N, 2.63. [g] $E^{5.8}$ +612.0 (c 0.2, CHCl₃).

(b) Synthesis of cis-1,2:3,4-diisopropylidene-5-C[acenaphthene-1"-onespiro[2".2']-(3'-ethoxycarbonyl)-1'-N-methylpyrrolidine]- α -D-galactopyranose (**6b**): isolated yield: 88% (0.430 g, 0.80 mmol) from 0.30 g of (2)-ethyl 6,7-dideoxy-1,2:3,4-di-O-isopropylidene- α -D-galacto-6-eno-octopyranuronate (0.91 mmol). Yellow solid. Mp = 128 °C. ¹H NMR (300 MHz, CDCl₃) δ = 0.86 (t, *J* = 7.2 Hz, 3H); 1.27 (s, 3H); 1.35 (s, 3H); 1.49 (s, 3H); 1.66 (s, 3H); 2.14 (s, 3H); 3.24 (d, *J* = 6.9 Hz, 1H); 3.49–3.55 (m, 1H); 3.58–3.62 (m, 2H); 3.75–3.86 (m, 1H); 3.93 (dd, *J* = 2.4, 7.5 Hz, 1H); 3.97–4.00 (m, 1H) 4.24–4.27 (m, 1H); 4.30–4.33 (m, 1H); 4.54 (dd, *J* = 2.4, 7.5 Hz, 1H); 5.55 (d, *J* = 5.1 Hz, 1H); 7.49 (d, *J* = 6.9 Hz, 1H); 7.57–7.62 (m, 1H); 7.70–7.75 (m, 1H); 7.85–7.92 (m, 2H); 8.10 (d, *J* = 8.1 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ = 12.7, 23.5, 24.1, 25.0, 25.1, 34.8, 36.6, 51.8, 57.0, 59.2, 67.4,

69.4, 69.9, 70.6, 77.7, 95.5, 107.8, 108.4, 120.1, 120.8, 124.1, 127.0, 127.2, 129.5, 129.9, 130.7, 135.6, 141.1, 169.6, 205.8. IR (KBr) ν_{max} : 1730, 1716 cm $^{-1}$. HRMS (EI) exact mass calcd for $C_{30}H_{35}NO_8$ 537.2363 (M⁺), found 537.2360. Anal. Calcd for $C_{30}H_{35}NO_8$: C, 67.02; H, 6.56; N, 2.61. Found: C, 67.10; H, 6.59; N, 2.60. $[\alpha]_{27}^{D.9}$ +537.6 (c 0.2, CHCl_3).

- 18. Grigg, R. Tetrahedron: Asymmetry 1995, 6, 2475-2486.
- Synthesis of trans-1,2:3,4-diisopropylidene-5-C[acenaphthene-1"-onespiro[2".2']-(3'-ethoxycarbonyl)-pyrrolizidine]-α-D-galactopyranose (7a): isolated yield: 86% (0.440 g, 0.78 mmol) from 0.30 g of (E)-ethyl 6.7-dideoxy-1,2:3,4-di-Oisopropylidene-α-D-galacto-6-eno-octopyranuronate (0.91 mmol). Yellow solid. Mp = 174 °C. ¹H NMR (300 MHz, CDCl₃) δ = 0.03 (L J = 7.2 Hz, 3H); 1.29 (s, 3H); 1.35 (s, 3H); 1.37 (s, 3H); 1.47 (s, 3H); 1.72-1.85 (m, 3H); 2.28-2.38 (m, 2H); 2.52-2.59 (m, 1H); 2.93-3.00 (m, 1H); 3.28 (q, J = 7.2 Hz, 2H); 4.01-4.06 (m, 1H); 4.07 (d, J = 11.4 Hz, 1H); 4.23-4.33 (m, 3H); 4.61 (dd, J = 2.1, 8.1 Hz, 1H); 5.61 (d, J = 5.1 Hz, 1H); 7.50 (d, J = 6.9 Hz, 1H); 7.62 (L, J = 8.1 Hz, 1H); 7.73 (t, J = 8.1 Hz, 1H); 7.85 (d, J = 8.4 Hz, 1H); 8.01 (d, J = 6.9 Hz, 1H); 8.09 (d, J = 8.4 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ = 10.9, 22.8, 23.3, 23.9, 24.1, 26.5, 32.0, 45.8, 45.9, 55.6, 58.0, 64.2, 64.5, 64.6, 69.0, 69.6, 71.6, 95.0, 106.6, 107.4, 119.8, 121.1, 123.3, 126.2, 126.3, 126.4, 128.8, 129.6, 130.7, 135.2, 169.0, 204.8. HRMS (E) exact mass calcd for C₃₂H₃₇NO₈ 564.2597 (M+H)⁺, found 564.2594. IR (KBr) v_{max}: 1730, 1716 cm⁻¹. Anal. Calcd for C₃₂H₃₇NO₈: C, 68.19; H, 6.62; N, 2.49. Found: C, 68.21; H, 6.69; N, 2.40. [x]₀^{27.7} +277.8 (c 0.2, CHCl₃).
 Synthesis of trans-1,2:3,4-diisopropylidene-5-C[spiro[2'.3"]oxindolo-3'-ethoxy-
- 20. Synthesis of trans-1,2:3,4-diisopropylidene-5-C[spiro]2'.3" Joxindolo-3' -ethoxy-carbonyl-1'-N-methylpyrrolidine]-α-D-galactopyranose (**9a**): isolated yield: 84% (0.380 g, 0.76 mmol) from 0.30 g of (*E*)-ethyl 6,7 dideoxy-1,2:3,4-di-O-isopropylidene-α-D-galacto-6-eno-octopyranuronate (0.91 mmol). Colorless solid. Mp = 130 °C. ¹H NMR (300 MHz, CDCl₃) δ = 0.79 (t, *J* = 7.2, 3H); 1.31 (s, 3H); 1.32 (s, 3H); 1.45 (s, 3H); 1.47 (s, 3H); 2.07 (s, 3H); 3.15 (t, *J* = 9 Hz, 1H); 3.21-3.31 (m, 1H); 3.57 (d, *J* = 9.0 Hz, 1H); 3.60-3.63 (m, 1H); 3.70-3.81 (m, 2H); 3.96 (d, *J* = 3.9 Hz, 1H); 4.25-4.28 (m, 1H); 4.31 (dd, *J* = 1.5, 5.1, 1H); 6.98 (t, *J* = 7.5 Hz, 1H); 7.17-7.24 (m, 2H); 8.73 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ = 13.5, 24.4, 25.1, 25.8, 35.4, 40.4, 54.3, 54.9, 60.4, 67.6, 70.6, 71.1, 72.9, 73.0, 96.7, 108.5, 109.2, 109.9, 122.3, 125.6, 126.8, 129.1, 141.3, 170.7, 179.1. IR (KBr) ν_{max} : 3296, 1730, 1710 cm⁻¹. HRMS (El) exact mass calch of C₂₆H₃₄N₂O₈: 63.2393 (M+H)⁺ found 503.2399. Anal. Calcd for C₂₆H₃₄N₂O₈: C, 62.14; H, 6.82; N, 5.57. Found: C, 62.19; H, 6.94; N, 5.65. [α]_D²⁸ 34.2 (c 0.2, CHCl₃). 21. The detailed X-ray crystallographic data (CCDC numbers for **9a** and **10b** are
- 21. The detailed X-ray crystallographic data (CCDC numbers for **9a** and **10b** are 763779 and 763778, respectively) are available from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge, CB2 1EZ, UK.